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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/595,454	04/20/2006	Gen Okiyama	SHM-16622	9943
	7590 03/17/201 L & CLARK LLP	0	EXAMINER	
38210 Glenn A	venue		TADAYYON ESLAMI, TABASSOM	
WILLOUGHBY, OH 44094-7808			ART UNIT	PAPER NUMBER
			1792	
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			03/17/2010	PAPER

# Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

		Application No.	Applicant(s)			
		10/595,454	OKIYAMA ET AL.			
Office Action Sum	mary	Examiner	Art Unit			
		TABASSOM TADAYYON ESLAMI	1792			
The MAILING DATE of this Period for Reply	communication app	ears on the cover sheet with the o	correspondence ad	ldress		
after SIX (6) MONTHS from the mailing date - If NO period for reply is specified above, the - Failure to reply within the set or extended pe	M THE MAILING DA ne provisions of 37 CFR 1.13 of this communication. maximum statutory period w riod for reply will, by statute, ree months after the mailing		N. mely filed the mailing date of this c ED (35 U.S.C. § 133).	·		
Status						
1) Responsive to communica	ion(s) filed on <u>19 No</u>	ovember 2009.				
2a)⊠ This action is <b>FINAL</b> .		action is non-final.				
,	<del>-</del>					
Disposition of Claims						
4) ☐ Claim(s) 1-8 is/are pending 4a) Of the above claim(s) is/are allow 6) ☐ Claim(s) 1-8 is/are rejected 7) ☐ Claim(s) 1-8 is/are object 8 ☐ Claim(s) are subject	is/are withdrav ved. l. cted to.					
Application Papers						
9)☐ The specification is objected	d to by the Examine	r.				
10) The drawing(s) filed on is/are: a) accepted or b) objected to by the Examiner.						
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).						
Replacement drawing sheet(s 11) The oath or declaration is o	-	on is required if the drawing(s) is ob aminer. Note the attached Office	-	* *		
Priority under 35 U.S.C. § 119						
<ul><li>2. Certified copies of th</li><li>3. Copies of the certifie application from the</li></ul>	one of: e priority documents e priority documents d copies of the prior International Bureau	s have been received. s have been received in Applicat ity documents have been receive	ion No ed in this National	Stage		
Attachment(s)		_				
<ol> <li>Notice of References Cited (PTO-892)</li> <li>Notice of Draftsperson's Patent Drawing</li> <li>Information Disclosure Statement(s) (Propage No(s)/Mail Date</li> </ol>		4)  Interview Summary Paper No(s)/Mail D 5)  Notice of Informal F 6)  Other:	ate			

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#### **DETAILED ACTION**

### Claim Rejections - 35 USC § 103

- 1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all abbasis rejections set forth in this Office action:
  - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 2. Claims 1- 8 are rejected under 35 U.S.C. 103(a) as being unpatentable over Osamu Sumiya et al (Japanese Patent Publication: 2003-229141, here after Sumiya), further in view of Kazuhito Hatoh et al (U. S. Patent Application: 2003/0143454, here after Hatoh), and Thomas Soczka Guth et al (U. S. Patent: 6632847, here after 847), and Tsutomu Seki et al (Japanese Patent: 408148152, here after Seki).

Claims 1 and 3 are rejected. Sumiya teaches a method of making electrode membrane assembly for a fuel cell by coating negative electrode diffusion layer (104) with a substrate layer (106) coating the substrate layer with negative electrode layer (101, catalyst layer), coating the electrode layer with ion ex-change layer(103, electrolyte layer) comprising to form electrolyte membrane [0011, 1112], coating the electrolyte membrane with positive electrode layer(102, catalytic layer), and coating the positive diffusion layer(105) with a substrate layer(107) to produce an electrode assembly membrane[fig. 9, 0005, abstract]. Sumiya does not teach coating the electrolyte membrane with positive electrode layer(102, catalytic layer), and superimposing on the other electrode layer a two layer body formed by coating the positive diffusion layer(105) with a substrate layer(107) to produce an electrode assembly membrane. Hotah teaches a method of making a fuel cell[ abstract ], where positive electrode layer(3 and 4) is laminated on

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electrolyte layer coated with catalytic layer(1 and 2)[fig.1, 0090]. In fact Hotah teaches forming the electrode layer separately and then bring it in contact with electrolyte layer assembly. Therefore, it would have been obvious to one of ordinary skill in the art at the time of invention was made to have a method of making electrode assembly as Sumiya teaches where the layer assembly that Sumyia teaches superimposed with positive electrode layer (105) and (107) as Hotah teaches, because Hotah teaches it is suitable method of making fuel cells. Although Sumiya does not clearly teaches applying the negative electrode layer before the substrate layer is dried, or applying the electrolyte layer before the substrate layer I dried, however Sumiya teaches for reducing the adhesion defects and improving adhesion between the layers, a coating solution has to applied to a layer before the layer is completely dried[0010]. Therefore, it would have been obvious to one of ordinary skill in the art at the time of invention was made to have a method of making electrode assembly as Sumiya and Hotah teach where each layer is coated on previous layer before the previous layer gets dried to increase the adhesion between the layers as Sumiya teaches. Sumiya also teaches drying the structure under no load [0015]. Sumiya does not teach the electrolyte(ion-exchange layer) comprising hydrocarbon solid polymer. 847 teaches a method of making a polymer ion change membrane for fuel cell [abstract], which comprising dissolving a solid polymer in a solvent comprising hydrogen and carbon [column 2 lines 32-36, column 7 lines 3-9] and coating the substrate with it. Therefore, it would have been obvious to one of ordinary skill in the art at the time of invention was made to have a method of making electrode assembly as Sumiya teaches where the electrolyte layer is taught by the method that 847 teaches, because 847 teaches a suitable electrolyte for making fuel cells. 847 also teaches drying the electrolyte layer at temperature of 80-140C[column 8 lines 60 to end, column 9, lines

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1-2] which is inherently less than the decomposition temperature of the polymer, otherwise the device would be useless. The do not teach placing electrode membrane assembly in vapor and removing the solvent from electrolyte membrane with . Seki teaches a method of making fuel cells where the electrode membrane is placed in water vapor and heat to remove the solvent to avoid forming crakes[abstract, constitution]. Therefore, it would have been obvious to one of ordinary skill in the art at the time of invention was made to have a method of making electrode assembly as Sumiya and 847 teach where the solvent is removed under vapor, because Seki teaches it helps to avoid forming cracks. Although Seki does not teach the drying temperature is less than the boiling temperature on the solvent, however since 847 teaches the solvent is N-methyl1-2-pyrrolidone [column 7 lines 3-9]. Therefore it is inherent the boiling point of N-methyl1-2-pyrrolidone (202 C) is higher than the drying temperature(80-140C).

Claim 2 is rejected for the same reason claim 1 is rejected. It is inherent that the solvent should be removed at lower temperature than the decomposition of the polymer, otherwise the device is useless.

Claim 4 is rejected, 847 teaches the solvent is N-methyll-2-pyrrolidone [column 7 lines 3-9]. Therefore, it would have been obvious to one of ordinary skill in the art at the time of invention was made to have a method of making electrode assembly as Sumiya Hotah, 847 and Seki teach where the solvent is N-methyll-2-pyrrolidone as 847 teaches, because 847 teaches it is suitable solvent for hydrocarbon polymers usable as electrolyte in fuel cells.

Claim 5 is rejected for the same reason claims 2 and 3 are rejected.

Claims 6-8 are rejected for the same reason claims 2-3, 5 and 4 are rejected.

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## Response to Arguments

3. Applicant's arguments filed 11/19/9 have been fully considered but they are not persuasive. The applicant argues that one ordinary skill in art would not combine Sumiya and Hatoh, however they are both teaching a method of fuel cells and some process done by Sumiya can be replaced by Hatoh. Furthermore selection of any order of performing process steps is prima facie obvious in the absence of new or unexpected results [MPEP. 2144.04]. The applicant argues Seki does not teach drying temperature less than the solvent boiling point, however since the solvent is taught by 847 and is N-methyl1-2-pyrrolidone [column 7 lines 3-9]. Therefore inherently the drying temperature (80-140C) is below the boiling point (202C).

#### Conclusion

4. **THIS ACTION IS MADE FINAL.** Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

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5. Any inquiry concerning this communication or earlier communications from the

examiner should be directed to TABASSOM TADAYYON ESLAMI whose telephone number

is (571)270-1885. The examiner can normally be reached on 7:30-5.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, Michael Cleveland can be reached on 571-272-1418. The fax phone number for the

organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent

Application Information Retrieval (PAIR) system. Status information for published applications

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like assistance from a USPTO Customer Service Representative or access to the automated

information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

Tabassom T. Tadayyon-Eslami

Examiner

Art Unit 1792

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Examiner, Art Unit 1792

/Michael Cleveland/

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